

**Faculty Research Grant 2014**

**Final Report**

**Water Soluble Hydrogenase Model**

Principal Investigator

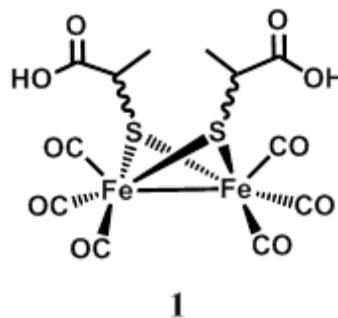
**CHARLES A. MEBI**

Department of Physical Sciences

## Statement of Problem

This project is aimed at preparing and studying the structure, electronic, and catalytic properties of an iron-carbonyl cluster coupled to thiolactate. The cluster is of interest as catalyst for the production of hydrogen, an environmentally benign alternative to fossil fuels. The development of efficient catalysts for hydrogen generation is an important step towards achieving the hydrogen economy.

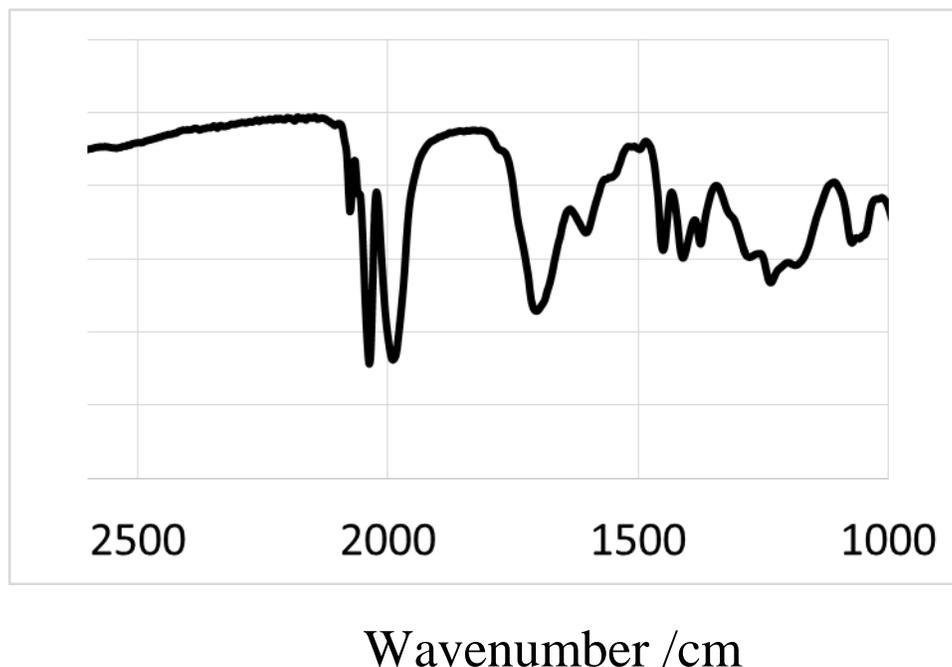
In the design of the catalyst (**1**), we incorporated thiolactate, a water soluble ligand to modulate the solubility property of the cluster. The characterization of this compound will be accomplished using spectroscopic techniques. This study will contribute to the fundamental knowledge on the design of effective catalysts for hydrogen production



## Brief Review of Research Procedure Utilized.

The proposed cluster was prepared following a single step procedure under nitrogen atmosphere using Schlenk line techniques. A mixture of thiolactate and triiron dodecacarbonyl was treated with THF solvent and the resultant solution refluxed for 30 minutes. A color change from green to red was observed. Removal of solvent followed by column chromatographic separation afforded the desired product as an orange oily substance. The structure and electronic properties of the compound were probed using infrared spectroscopy, and examined as electrocatalyst for the generation of hydrogen by an electrochemical technique (cyclic voltammetry).

Figure 1: IR spectrum of desired compound in DCM



#### SUMMARY OF FINDINGS

The desired compound has been successfully prepared, isolated and characterized by IR spectroscopy. IR spectrum of the compound shows peaks between 2000 and 2100  $\text{cm}^{-1}$  ascribed to terminal metal carbonyls. Partial results were presented at the ACS Regional meeting in Pocatello, Idaho, June 25-29, 2015.

#### CONCLUSIONS AND RECOMMENDATIONS

We have prepared the new iron-cluster, compound 1. Further characterization is being carried out. We gratefully acknowledge partial support for this project from Arkansas Tech University Faculty Research Grant (2012-2013).

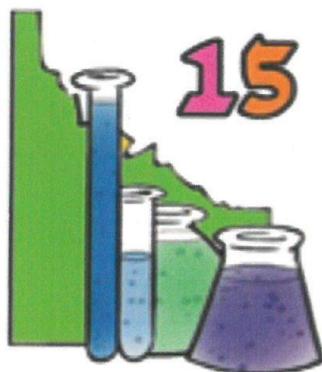
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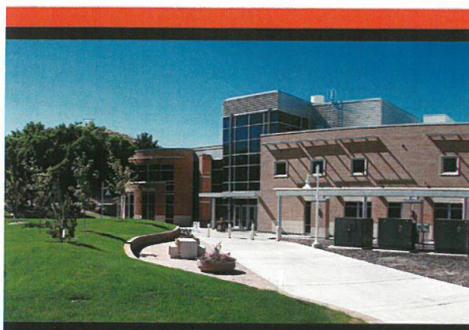
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# The 70<sup>th</sup> Northwest Regional Meeting of the American Chemical Society



**N O R M**



**June 21 – 24, 2015**  
On the campus of  
**Idaho State University**  
Pocatello, Idaho

## TUESDAY EVENING

Pond Student Union  
Exhibit Room

### Emerging Trends in Actinide and Lanthanide Separations

B. Mincher, *Organizer*

5:00 - 7:00

93. Adsorption of uranium using an aminophosphonic acid functionalized material. **R.L. Tsosie**, E. Rosenberg

94. Systematic study of N,N,N',N'-tetra-2-ethylhexyl diglycolamide (T2EHDGA) solvent extraction behavior. **E.L. Campbell**, T.G. Levitskaia

Pond Student Union  
Exhibit Room

### Inorganic Chemistry

Cosponsored by INOR<sup>‡</sup>  
J. J. Pak, *Organizer*

5:00 - 7:00

95. Hydrogen evolution from neutral water catalyzed by hydrogenobacter thermophilus cobalt-cytochrome *c*. **B. Kandemir**, T.A. Ruberu, R. Eisenberg, K. Bren

96. Varying the elemental ratios in Cu<sub>2</sub>ZnSnS<sub>4</sub> nanoparticles. **K.V. Nielson**, J. Chavez, J.J. Pak

97. Preparation and study of quaternary quantum dots. **T. Huntsinger**, **J. Chavez**, J.J. Pak

98. Toxicology studies of semiconductor nanoparticles. **D.R. Walker**, J.J. Pak

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99. Bio-inspired iron-based hydrogen producing catalysts. **C.A. Mebi**, A.L. Haley, J. Randall, S.L. Moran

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100. Synthesis of model compounds that mimic the primary and secondary coordination sphere of carbonic anhydrase. **J. Elsberg**, E.C. Brown